metal-organic papers

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Key indicators

Single-crystal X-ray study T = 150 KMean $\sigma(C-C) = 0.018 \text{ Å}$ R factor = 0.064 wR factor = 0.127 Data-to-parameter ratio = 14.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Bis[di-η⁵-cyclopentadienylcobalt(I)] dicarbonyltrichloro(chlorodiphenylsilyl)ruthenium(II)

The asymmetric unit of the title compound, $[Co(C_5H_5)_2]_2$ -[RuCl₃(C₁₂H₁₀ClSi)(CO)₂], consists of a discrete ClPh₂Si-Ru(CO)₂Cl₃ dianion and two $(\eta$ -C₅H₅)₂Co cations. The dianion has octahedral coordination geometry, with the three Cl atoms in *cis* positions. In addition, two carbonyl groups (*cis*) and a Ph₂ClSi group (bonded through Si) complete the coordination. The Ru–Si bond length is 2.362 (3) Å and the longest Ru–Cl bond [2.566 (2) Å] is *trans* to the Si atom. In one cobaltocene cation, the cyclopentadienyl (*Cp*) rings are staggered, while in the other, the *Cp* rings are almost eclipsed.

Comment

Metal complexes with functional groups are of potential interest as precursors to metallopolymers. As part of our work in this area (Nguyen *et al.*, 1999), we report the characterization of an unexpected product, (I).



Experimental

The reaction between two equivalents of $(\eta$ -C₅H₅)₂Co and [ClPh₂SiRu(CO)₄]₂ was carried out at 298 K in tetrahydrofuran under an atmosphere of N₂. After *ca* 16 h, the solvent was removed under high vacuum and CD₂Cl₂ was added to the residue. A vigorous reaction was observed with evolution of gas. Red–orange crystals of (I) were obtained upon slow room-temperature concentration of the resultant product mixture under an N₂ atmosphere.

Crystal data	
$\begin{split} & [\text{Co}(\text{C}_5\text{H}_5)_2]_{2^-} \\ & [\text{RuCl}_3(\text{C}_{12}\text{H}_{10}\text{ClSi})(\text{CO})_2] \\ & M_r = 859.40 \\ & \text{Orthorhombic}, P2_12_12_1 \\ & a = 9.7462 \ (5) \text{ Å} \\ & b = 10.2277 \ (6) \text{ Å} \\ & c = 34.099 \ (2) \text{ Å} \\ & V = 3399.0 \ (3) \text{ Å}^3 \\ & Z = 4 \end{split}$	$D_x = 1.679 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 3060 reflections $\theta = 2.6-25.0^{\circ}$ $\mu = 1.78 \text{ mm}^{-1}$ T = 150 (1) K Plate, dark orange $0.20 \times 0.20 \times 0.02 \text{ mm}$
Data collection	
Nonius KappaCCD diffractometer φ scans and ω scans with κ offsets Absorption correction: multi-scan (<i>DENZO-SMN</i> ; Otwinowski & Minor, 1997) $T_{min} = 0.717, T_{max} = 0.965$ 12 795 measured reflections	5652 independent reflections 3558 reflections with $I > 2\sigma(I)$ $R_{int} = 0.109$ $\theta_{max} = 25.0^{\circ}$ $h = -11 \rightarrow 11$ $k = -12 \rightarrow 12$ $l = -37 \rightarrow 36$

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Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_z^2) + (0.0371P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.064$	where $P = (F_0^2 + 2F_c^2)/3$
$wR(F^2) = 0.128$	$(\Delta/\sigma)_{\rm max} = 0.001$
S = 1.01	$\Delta \rho_{\rm max} = 0.84 \ {\rm e} \ {\rm \AA}^{-3}$
5652 reflections	$\Delta \rho_{\rm min} = -0.54 \ {\rm e} \ {\rm \AA}^{-3}$
397 parameters	Absolute structure: (Flack, 1983),
H-atom parameters constrained	2319 Friedel pairs
-	Flack parameter $= 0.03$ (3)

Table 1

Selected geometric parameters (Å, °).

Ru1-Si1	2.362 (3)	Ru1-Cl1	2.566 (2)
Ru1-Cl2	2.456 (3)	Cl4-Si1	2.134 (4)
Ru1-Cl3	2.462 (3)		
Si1-Ru1-Cl1	173.57 (10)	Cl4-Si1-Ru1	111.99 (15)
C1-Ru1-Si1-Cl4	-11.3 (4)		

All H atoms were included in calculated positions with distances of 0.95 Å for phenyl C–H and 1.00 Å for cyclopentadienyl C–H. In the refinement, H atoms were included in a riding-motion approximation, with $U_{\rm iso} = 1.2U_{\rm eq}$ of the carrier atom.

Data collection: *COLLECT* (Nonius, 1997–2001); cell refinement: *DENZO-SMN* (Otwinowski & Minor, 1997); data reduction: *DENZO-SMN*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2001); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.



Figure 1

View of molecule (I), with displacement ellipsoids drawn at the 30% probability level.

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